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5-tert-Butyl-4-bromo-1,2-dihydro-1*H*-pyrazol-3(2*H*)-one monohydrate

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Key indicators

Single-crystal X-ray study $T=150~\mathrm{K}$ Mean $\sigma(\mathrm{C-C})=0.004~\mathrm{\mathring{A}}$ R factor = 0.030 wR factor = 0.069 Data-to-parameter ratio = 15.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The structure of the title compound, $C_7H_{11}BrN_2O\cdot H_2O$, exhibits an elaborate hydrogen-bonding network involving pyrazole $N-H\cdot\cdot\cdot O$ dimers and two other hydrogen-bonding motifs, both including water molecules. One motif is a distorted hexagonal $R_3^5(11)$ graph set, while the other is a distorted octagonal boat conformation $R_6^4(14)$ graph set.

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Comment

In a series of studies on the preparation and hydrogen-bonding properties of 3,4,5-tri-substituted pyrazoles, we recently characterized the structure of 5-tert-butyl-4-nitro-1H-pyrazol-3-ol (Lynch & McClenaghan, 2005). We report here the structure of the title compound, (I). Similar to 5-tert-butyl-4-nitro-1H-pyrazol-3-ol, compound (I) originated from 3,5-di-tert-butylpyrazole. Compound (I) was prepared by reacting 3,5-di-tert-butylpyrazole with bromine in chloroform solution at room temperature. In these reactions, 3,5-di-t-butylpyrazole is attacked by either nitric acid (as in the case of 5-tert-butyl-4-nitro-1H-pyrazol-3-ol) or bromine to form the onium species, which then displaces one tert-butyl group. The subsequent vacant position is then filled by an OH group that, in the case of (I), tautomerizes to form the pyrazolone.

$$\begin{array}{c} & & \\$$

In the structure of (I) (Fig. 1), all strong hydrogen-bonding components are involved in the hydrogen-bonding network. The hydrogen-bonding geometry for this structure is listed in Table 1. The fourfold symmetry in (I) arises because of the unique hydrogen-bonded motif that is formed via contributions from eight pyrazole molecules and four water molecules. Each pyrazole molecule forms a centrosymmetric $R_2^2(8)$ graph set (Etter, 1990) dimer via N1-H \cdots O5 interactions, at (x, y, z) and $(-x + \frac{3}{2}, -y + \frac{1}{2}, -z + \frac{1}{2})$, centred at $(\frac{3}{4}, \frac{1}{4}, \frac{1}{4})$. The N2/H group associates with O1W, at (x, y, z) and $(y + \frac{1}{4}, -x + \frac{5}{4}, z + \frac{1}{4})$. O1W, at (x, y, z), associates with two O5 atoms, one at (x, y, z)and the other at $\left(-y + \frac{5}{4}, x - \frac{3}{4}, -z + \frac{1}{4}\right)$. Thus, each O5 atom is involved in a four-centre hydrogen-bonding association. For O5, at (x, y, z), the three non-H-atom contacts are O1W at (x, y, z)(y, z), O1I at $(y + \frac{3}{4}, -x + \frac{5}{4}, -z + \frac{1}{4})$ and N1 at $(-x + \frac{3}{2}, -y + \frac{1}{2}, -y + \frac{1}{2})$ $-z + \frac{1}{2}$) (Fig. 2). Three pyrazole molecules, at (x, y, z), $(-x + \frac{3}{2})$

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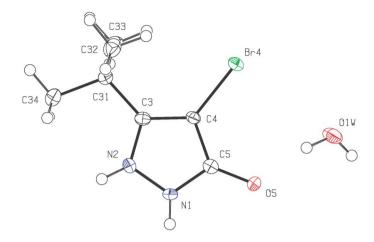


Figure 1 Molecular configuration and atom-numbering scheme for (I). Displacement ellipsoids are drawn at the 50% probability level and H atoms are drawn as spheres of arbitrary radii.

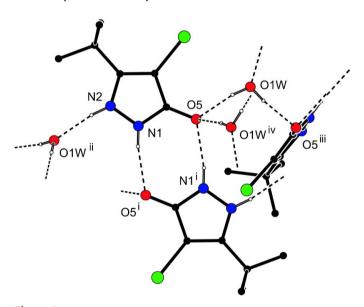


Figure 2

Hydrogen-bonding environment for (I), at (x, y, z), showing the centrosymmetric $R_2^2(8)$ N1-H \cdots O5 dimer, the two hydrogen-bonding associations from O1W, and the four-centre hydrogen-bonding association involving O5. For clarity, H atoms not involved in the hydrogen-bonding interactions have been omitted. Hydrogen bonds are shown as dashed lines. [Symmetry codes: (i) $-x+\frac{3}{2},-y+\frac{1}{2},-z+\frac{1}{2}$; (ii) $y+\frac{1}{4},-x+\frac{5}{4},z+\frac{1}{4}$; (iii) $-y+\frac{5}{4},x-\frac{3}{4},-z+\frac{1}{4}$; (iv) $y+\frac{3}{4},-z+\frac{1}{4}$, $-z+\frac{1}{4}$.

 $-y+\frac{1}{2},-z+\frac{1}{2})$ and $(-y+\frac{5}{4},x-\frac{3}{4},-z+\frac{1}{4})$, and two water molecules, at (x,y,z) and $(-y+\frac{5}{4},x-\frac{3}{4},-z+\frac{1}{4})$, form a distorted hexagonal hydrogen-bonding motif [graph set $R_3^5(11)$], adjoining the N1-H···O5 dimer, fused *via* the same interaction (Fig. 3). The hexagonal motifs are also fused with each other *via* the O1W-H···O5 interaction at (x,y,z). The resulting arrangement also creates a distorted octagonal boat conformation hydrogen-bonding motif [graph set $R_6^4(14)$] involving four pyrazole groups, at (x,y,z), $(-x+\frac{3}{2},-y+\frac{1}{2},-z+\frac{1}{2})$, $(x-\frac{1}{2},y,-z+\frac{1}{2})$ and $(1-x,-y+\frac{1}{2},z)$, and two water molecules, at $(y+\frac{1}{4},-x+\frac{5}{4},z+\frac{1}{4})$ and $(-y+\frac{3}{4},x-\frac{3}{4},z+\frac{1}{4})$

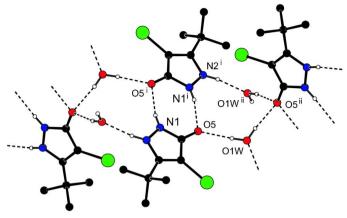


Figure 3 Part of the structure of (I), at (x, y, z), showing the distorted $R_3^5(11)$ hexagonal motif, and its position with respect to the N1-H \cdots O5 dimer. For clarity, H atoms not involved in the hydrogen-bonding interactions have been omitted. Hydrogen bonds are shown as dashed lines. [Symmetry codes: (i) $-x + \frac{3}{2}$, $-y + \frac{1}{2}$, $-z + \frac{1}{2}$; (ii) $-y + \frac{5}{4}$, $x - \frac{3}{4}$, $z - \frac{1}{4}$.]

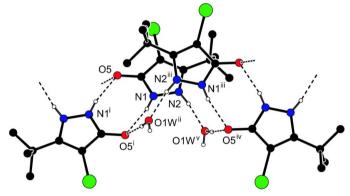


Figure 4

Part of the structure of (I), at (x, y, z), showing the distorted $R_6^4(14)$ octagonal boat motif and its position with respect to the N1-H···O5 dimer. For clarity, H atoms not involved in the hydrogen-bonding interactions have been omitted. Hydrogen bonds are shown as dashed lines. [Symmetry codes: (i) $-x + \frac{3}{2}, -y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-y + \frac{3}{4}, x - \frac{3}{4}, z + \frac{1}{4}$; (iii) $1 - x, -y + \frac{1}{2}, z$; (iv) $x - \frac{1}{2}, y, -z + \frac{1}{2}$; (v) $y + \frac{1}{4}, -x + \frac{5}{4}, z + \frac{1}{4}$.]

(Fig. 4). A stereoview of the unit cell contents of (I) is shown in Fig. 5. The Br atom does not contribute to the hydrogen-bonding network; atom Br4 is 3.469 (3) Å from O1W, and 3.412 (3) Å from N2($-y + \frac{5}{4}$, $x - \frac{1}{4}$, $z - \frac{1}{4}$).

Experimental

The title compound was obtained from Key Organics Ltd, and crystals were grown from an ethanol solution.

Crystal data

 $C_7H_{11} Br N_2 O \cdot H_2 O$ Mo $K\alpha$ radiation $M_r = 237.10$ Cell parameters from 2331 Tetragonal, $I4_1/a$ reflections a = 13.6840 (4) Å $\theta = 2.9-27.5^\circ$ c = 21.4734 (8) Å $\mu = 4.06 \text{ mm}^{-1}$ V = 4020.9 (2) Å³ T = 150 (2) K Z = 16 Prism, colourless $D_x = 1.567 \text{ Mg m}^{-3}$ 0.36 × 0.27 × 0.20 mm

Data collection

Nonius KappaCCD diffractometer φ and ω scans Absorption correction: multi-scan (SADABS; Sheldrick, 2003) $T_{\min} = 0.288$, $T_{\max} = 0.444$ 12 489 measured reflections 1977 independent reflections

1633 reflections with
$$I > 2\sigma(I)$$

 $R_{\rm int} = 0.069$
 $\theta_{\rm max} = 26.0^{\circ}$
 $h = -16 \rightarrow 16$
 $k = -15 \rightarrow 14$
 $l = -19 \rightarrow 26$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.030$ $wR(F^2) = 0.069$ S = 1.021977 reflections 124 parameters H atoms treated by a mixture of independent and constrained refinement

$$\begin{split} w &= 1/[\sigma^2(F_{\rm o}^2) + (0.0247P)^2 \\ &+ 7.0406P] \\ \text{where } P &= (F_{\rm o}^2 + 2F_{\rm c}^2)/3 \\ (\Delta/\sigma)_{\rm max} &< 0.001 \\ \Delta\rho_{\rm max} &= 0.30 \text{ e Å}^{-3} \\ \Delta\rho_{\rm min} &= -0.64 \text{ e Å}^{-3} \end{split}$$

$D-H\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-H\cdots A$
$ \begin{array}{c} N1 - H1 \cdots O5^{i} \\ N2 - H2 \cdots O1W^{ii} \\ O1W - H1W \cdots O5 \end{array} $	0.82 (3) 0.85 (3) 0.83 (2)	1.97 (3) 1.80 (3) 1.91 (2)	2.773 (3) 2.646 (3) 2.733 (3)	168 (3) 171 (3) 169 (3)
$\frac{\text{O1}W - \text{H2}W \cdot \cdot \cdot \text{O5}^{\text{III}}}{\text{Symmetry codes: (i)}}$	$0.83(2)$ $-x + \frac{3}{2}, -v$	$\frac{1.91 (2)}{+\frac{1}{2}, -z + \frac{1}{2}}$	2.739 (3) (ii) $y + \frac{1}{4}, -x + \frac{1}{4}$	$\frac{173 (3)}{\frac{5}{4}, z + \frac{1}{4}; \text{ (iii)}}$
$-y + \frac{5}{4}, x - \frac{3}{4}, -z + \frac{1}{4}.$, 2, ,	. 2, ~ . 2,	()) 1 4,	4, ~ , 4, ()

All *tert*-butyl H atoms were included in the refinement at calculated positions, in the riding-model approximation, with C–H distances of 0.98 Å. All NH H atoms involved in the hydrogenbonding associations (Table 1) were located in Fourier syntheses and positional parameters were refined. The water H atoms were located and were refined with O–H distance restraints of 0.83 (2) Å and $H\cdots H$ restraints of 1.40 (2) Å. The isotropic displacement parameters for all H atoms were set equal to $1.25U_{\rm eq}$ of the carrier atom.

Data collection: COLLECT (Hooft, 1998); cell refinement: DENZO (Otwinowski & Minor, 1997) and COLLECT; data reduc-

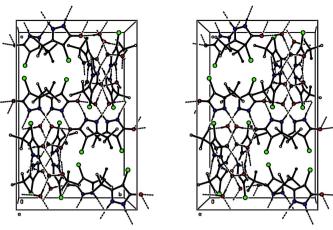


Figure 5
Stereoview of the unit cell contents of (I).

tion: *DENZO* and *COLLECT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON97* (Spek, 1997); software used to prepare material for publication: *SHELXL97*.

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